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rate laws and determining reaction order. Use each of the terms below to complete the statements ...

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 2 11 In your textbook, read about significant figures. Use each of the terms below just once to complete the statements. 7. The digits that are reported in an answer are called . 8. The numeral 9.66 has three significant figures, two

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known figures and one figure.

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- Chapter 3 13 Matter—Properties and Changes Matter—Properties and Changes Section 3.1 Properties of Matter In your textbook, read about physical properties and chemical properties of matter. Use each of the terms below just

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once to complete the passage.

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about how binary compounds and acids are named from their  
formulas.

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- Chapter 2 7 Data Analysis Data Analysis Section 2.1 Units of Measurement In your textbook, read about SI units. Complete the following table. In your textbook, read about base units and derived units. For each SI unit in Column A, write the letter of the matching item from Column B.

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- Chapter 10 59 Section 10.3 Reactions in Aqueous Solutions In

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your textbook, read about aqueous solutions, reactions that form precipitates, reactions that form water, and reactions that form gases.

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mmching item in Column B. Column A 1. The rays and particles that are emitted by a radioactive material

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- Chapter 21 123 In your textbook, read about calculating cell electrochemical potential. Use the table of standard reduction potentials below to answer the following questions. 20. Suppose you have two voltaic cells whose half-cells are represented by the following

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STUDY GUIDE FOR CONTENT MASTERY Section 6.3 Periodic Trends In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured directly because the electron cloud surrounding the nucleus does not have a clearly defined c. outer edge.

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