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General Chemistry. The eleventh edition was carefully reviewed with an eye toward strengthening the content available in OWLv2, end-of-chapter questions, and updating the presentation.

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Mass of tablet + mass of acid solution = mass of carbon dioxide + mass of solution

Substituting, you obtain $0.853 \text{ g} + 56.519 \text{ g} = \text{mass of carbon dioxide} + 57.152 \text{ g}$ Mass of carbon dioxide = $0.853 \text{ g} + 56.519 \text{ g} - 57.152 \text{ g} = 0.220 \text{ g}$ Thus,...

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