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ANSWERS To TOPIC 6 Review Questions 1. 2 2- 1 3- 2 CHEMISTRY Answm KIY 31 37. Diagrams of the molecules should be like the 35. Decrease the temperature to 35°C, or add 25 g ammonia molecule in Figure 6—10 on page 85. more solute.

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Answer Key For Chemistry The Physical Setting 2019

PHYSICAL SETTING CHEMISTRY Friday, January 25, 2019 — 9:15 a.m. to 12:15 p.m., only This is a test of your knowledge of chemistry. Use that knowledge to answer all questions in this examination. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

Physical Setting/Chemistry Core Curriculum

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Mg²⁺ and Al³⁺ are isoelectronic and have the electron configuration of Ne. S²⁻, Cl⁻, and K are also isoelectronic and have the electron configuration of Ar. Ba²⁺ has the electron configuration of Xe. 13 Ge and AS Group 1 metals all become ions with a 1+ charge. Solid The valence electrons of Group 2 are mobile.

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PHYSICAL SETTING CHEMISTRY Wednesday, January 28, 2015 — 1:15 to 4:15 p.m., only This is a test of your knowledge of chemistry. Use that knowledge to answer all questions in this examination. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry. You are to answer all questions in all

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Physical Setting/Chemistry Regents Examinations

CHEMISTRY ANSWER KEY 27 Na Mg Al 1s22s22p63s23p1 Si 1s22s22p63s23p2 P S 1s22s22p63s23p4 Ar 1s22s22p63s23p6 1 1 4 2 59. 4 62. 2 65. 2 68. 2 The mixture contains at least three parts. One solid remained on the filter paper. The other two parts consist of a liquid with a solid dissolved in it. If the calcium and sulfur can be separated by

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PHYSICAL SETTING CHEMISTRY - JMAP

to the Physical Setting/Chemistry Regents Examination, students must have successfully completed 1200 minutes of laboratory experience with satisfactory reports on file. Because of the strong emphasis on student development of laboratory skills, a minimum of 280 minutes per week of class and laboratory time is recommended.

Reviewing Chemistry: The Physical Setting

PHYSICAL SETTING CHEMISTRY Thursday, January 25, 2018 — 9:15 a.m. to 12:15 p.m., only This is a test of your knowledge of chemistry. Use that knowledge to answer all questions in this examination. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

Answers to Topic 5 Periodic Table from Review Book

Reference Tables for Physical Setting/Chemistry – 2011 Edition 5 Table I Heats of Reaction at 101.3 kPa and 298 K Table J Activity Series** Reaction ΔH (kJ)* CH 4 (g) + 2O (g) CO(g) + 2H O() –890.4 C 3 H 8 (g) + 5O (g) 3CO(g) + 4H O() –2219.2 2C 8 H 18 () + 25O (g) 16CO(g) + 18H O() –10943 2CH 3 OH() + 3O (g) 2CO(g) + 4H O() –1452 C 2 H 5 OH() + 3O 2 (g) 2CO 2